

Calcium (Ca): most abundant mineral in body.

• blood  $\text{Ca}^{2+} \Rightarrow 9-11 \text{ mg/dL}$

Dietary Source: milk & its products, millets, ragi

• Phytates (present in cereals) & oxalates (in green leafy vegetables) inhibit Ca absorption

RDA: Adult: 0.8 gm

Children: 0.8-1.2 gm

Pregnancy & Lactation: 1.5 gm

Absorption: absorbed in duodenum against conc. gradient

$\rightarrow$  absorption requires Calcium Binding Protein (CBP)

Factors facilitating Ca absorption:

- Calcitriol (active form of vit. D)  $\Rightarrow$  induces CBP synthesis
- PTH (promotes synthesis of calcitriol)
- Lactose
- Lysine, Arginine
- Gastric Acidity (low pH)

Factors decreasing Ca absorption:

- Phytates (in cereals)
- oxalates (in green leafy vegetables)
- fatty acids
- high phosphate content in food
- alkaline condition

Chronic renal failure/Nephrosis Causes Decreased  $\text{Ca}^{2+}$  Absorption:

- Kidney is required for formation of Calcitriol required for CBP synthesis
- can be treated by administration of calcitriol.

Steatorrhea decreases Ca absorption:

- steatorrhea occurs due to defective digestion & absorption of fat
- unabsorbed fatty acids inhibit Ca absorption

Calcium: Phosphate Ratio in Diet:

- Maximum ratio = 1:2 to 2:1
- Ideal ratio = 1:1

Storage: bones, teeth, muscles

Excretion: 500 mg/day in urine

### Functions of Ca:

- Bones & teeth formation: Ca gives hardness & strength
- Blood Coagulation
- Muscle contraction - for excitation & contraction of muscle fibres
- Transmission of nerve impulses - in pre & post-synaptic region.
- Neuromuscular excitability - Ca decreases excitability
- Hormone Action - secondary messenger to some hormones  
eg - glucagon, adrenalin, vasopressin
- Release of hormones - insulin, PTH, calcitonin

### Dietary Deficiency:

Children - impaired bone growth

- ricket like symptoms
- early occurrence of osteoporosis

Adults - osteoporosis

- hyperexcitable state of nerves & muscles

Toxicity: unlikely to reach toxic amounts

- increased risk of kidney stones
- decreased absorption of other minerals (Fe, Zn, P, Mg, etc.)

Calcium Homeostasis: blood Ca  $\Rightarrow$  9-11 mg/dL

Vitamin D: hypercalcemic

- sites of action = intestine, kidney, bones

Intestine: Calcitriol induces CBP synthesis in mucosa

Kidney: increases calcium reabsorption

Bones: enhances osteoblastic activity, calcification (causes bone resorption during <sup>hypo</sup> calcemia)

Parathyroid Hormone (PTH): hypercalcemic

Intestine: PTH stimulates production of Calcitriol  $\Rightarrow$  increased intestinal absorption of Ca

Kidney: increased Ca reabsorption

Bone: demineralization (resorption) of bones (mainly during Ca deficiency)

Calcitonin: hypocalcemic

Kidney: inhibits Ca reabsorption by renal tubules

Bone: inhibits bone resorption by increasing activity of osteoblasts

$PO_4^{2-}$  levels: Hypophosphatemia enhances hydroxylation of vit. D in kidneys to form Calcitriol  $\Rightarrow$  hypercalcemic effect

### • Activation of some enzymes

$\rightarrow$  As Ca-calmodulin complex - Calmodulin is a CBP;

- it can bind with 4  $Ca^{2+}$

eg:- Adenylate cyclase, Ca dependent protein kinase

$\rightarrow$  Direct action - pancreatic lipase, rennin

• Promotes & controls membrane integrity & cell permeability.

• Cell-to-Cell Contact

Hypocalcemia:  $< 9 \text{ mg/dL}$

Causes - vit. D deficiency

- hypoparathyroidism

- accidental surgical removal of parathyroid glands (during thyroidectomy)

- dietary deficiency

- Steatorrhea (due to accumulation of fatty acids, Ca absorption is inhibited)

- Chronic renal disease (impairment of Calcitriol formation)

Manifestations: increased neuromuscular excitability (mild tremors, tetany)

(Numbness of extremities, tightness, muscle spasms & cramps, convulsions, etc)

Positive Chvostek's sign & Trousseau's sign

Hypercalcemia:  $> 11$  mg/dL

Causes - hypervitaminosis D

- hyperparathyroidism

Symptoms - increased excretion of urinary Ca leading to renal calculi

- bone deformities

- ectopic calcification of urinary bladder, renal tissue, pancreas

- anorexia

- muscle weakness

- depression

Phosphorus (P): adult body has about 1 kg P.

Sources: milk, eggs, cereals

- Phytates (in plant sources) inhibit P absorption

RDA: Adults: 800 mg

Children: 1000 mg

Lactation & pregnancy: 1200 mg

Absorption: mainly from jejunum

↳ increased by Calcitriol & PTH

Storage: bones, teeth, muscles

Excretion: 500 mg/day

Functions:

- Formation of bones & teeth
- Production of high energy phosphates - ATP, GTP (for energy)
- Synthesis of nucleoside enzymes - NAD, NADP
- Nucleic acid synthesis in DNA & RNA. (phosphodiester bonds form backbone)
- Formation of phosphate esters → glucose-6-phosphate, etc.
- Enzyme activation (reversible covalent modification by phosphorylation & dephosphorylation)
- Buffer system in blood & cells.

Dietary deficiency: Rare

- Causes - vit. D deficiency  
- consumption of large amounts of antacid
- may cause osteomalacia, growth retardation, etc.

Toxicity: unlikely

↳ may develop bone resorption

Phosphorus Homeostasis: normal serum level = 3-4 mg/dl in adults

children = 5-6 mg/dL

(Haemolysed blood specimens are not used for P estimation)

∴ lysed RBCs release phosphate into plasma giving false higher value)

Vit. D: causes release of P from bones (by activating alkaline phosphatase)

PTH: decreases P reabsorption from renal tubules.

Hypophosphatemia: < 3 mg/dL

Causes: vit. D deficiency, hyperparathyroidism

Hypophosphatemia: > 4 mg/dL

Causes: hypervitaminosis D, hyperparathyroidism, renal failure

# Magnesium (Mg):

- 25g in body
- normal plasma level: 2-3 mg/dL

Sources: milk, green leafy vegetables, sea foods, cereals

RDA: Adult ♂ = 350mg

Adult ♀ = 300mg

Absorption: increased Ca & P decrease Mg absorption

↳ absorbed with the help of a specific carrier

Function: • Constituent of bone & teeth

- Cofactor for enzymes: hexokinase, fructokinase, PFK, enolase, ALA synthase, alkaline phosphatase
- Proper neuromuscular function: decreases neuromuscular excitability  
↳ along with Ca, Mg acts as a relaxant during activity, after contraction
- Insulin Sensitivity: Mg deficiency leads to decreased insulin dependent uptake of glucose
- Activates amino acids for protein synthesis
- Synthesis & maintenance of genetic material

Dietary deficiency: rare

→ manifests only in combination with protein-energy malnutrition (PEM), alcoholism, persistent diarrhoea.

Symptoms: muscle weakness, cramps, hypertension, etc.

Toxicity: unlikely

Cause: renal insufficiency being treated with supplementation

Symptoms: nausea, vomiting, diarrhoea.

Cu: Total body Cu = 100mg.

Sources: liver, fish, meat (milk is a poor source)

RDA: 2-3 mg

Absorption: from duodenum

↳ transport protein = metallothionein

→ phytates, zinc, molybdenum decrease Cu uptake

Functions: Formation of many enzymes - ALA synthase, cytochrome oxidase, dopamine oxidase

- Mobilisation of Iron -  $Fe^{2+} \xrightarrow{\text{ceruloplasmin (Cu}^{2+})} Fe^{3+}$
- Synthesis of Hb - Cu is a constituent of ALA synthase
- Formation of Collagen, Elastin - constituent of lysyl oxidase
- Synthesis of melanin - constituent of tyrosinase
- Catecholamine synthesis - constituent of dopamine  $\beta$ -hydroxylase
- Antioxidant function - as a component of SOD (superoxide dismutase)
- Role in ETC - constituent of complex IV - cyt. oxidase

Dietary Deficiency: microcytic anemia ( $\because$  Cu is needed for mobilisation & utilisation of Fe)

↳ defective collagen formation (osteoporosis, fragile capillaries)

Toxicity: by contamination of food cooked in copper vessels

↳ vomiting, diarrhoea, etc.

Wilson's Disease / Hepatolenticular Degeneration:

→ autosomal recessive disease

→ defect in ceruloplasmin synthesis

(defect in hepatic excretion of Cu into bile & renal absorption of Cu)

Manifestation: low blood Cu

→ Excessive deposition of Cu in liver & brain ⇒ cirrhosis, neurological disorders

→ deposition in kidney ⇒ renal damage

→ deposition in cornea ⇒ Kayser Fleischer ring (at margin of cornea)

Treatment: penicillamine (Cu chelating agent)

Menke's Disease (Kinky/Steely Hair Syndrome):

→ X-linked disorder

→ defect in intestinal absorption of Cu (Cu may be trapped by metallothionein in intesti

Symptoms: decreased plasma & urine Cu.

→ anemia → growth failure

→ mental retardation

→ depigmentation of skin & hair (kinky/steely hair)

→ usually fatal in infancy.

Fe: Total body iron = 3-5 g

Sources: Rich - organ meat, jaggery

good - leaf vegetables, fish

poor - milk, polished rice, wheat

→ Phytates, oxalates, phosphates inhibit Fe absorption

RDA: Man & post-menopausal woman: 10-20 mg

Pre-menopausal women: 20-40 mg

children: 15-30 mg

Pregnancy & lactation: 20-40 mg

Absorption: upper part of duodenum mainly

Factors influencing Fe Absorption -

- Amount of iron - directly proportional
- Chemical form of iron: absorbed in  $Fe^{2+}$  form
- Physical form: heme iron is better absorbed than non-heme iron.

Dietary iron

- Heme iron: absorbed into intestinal mucosal cells
- Non-heme iron: HCl in stomach releases Fe from non-heme proteins; this iron is in  $Fe^{3+}$  form;  $\therefore Fe^{3+}$  is reduced to  $Fe^{2+}$  by vitamin C, glutathione, HCl & then absorbed.

Factors stimulating Fe absorption:

- HCl:  $Fe^{3+} \rightarrow Fe^{2+} \rightarrow$  absorption
- vit. C, cysteine, glutathione:  $Fe^{3+} \rightarrow Fe^{2+} \rightarrow$  absorption
- vit. A, amino acids: stimulate Fe absorption by forming soluble salts with Fe
- Calcium: forms complexes with iron inhibitors like oxalates, phosphates.

Factors inhibiting Fe Absorption:

- Phytates, oxalates, phosphates: form insoluble complexes with Fe
- Fatty acids
- Tannic acid of tea

## Regulation of Fe Absorption:

**Mucosal Block Theory:** Fe homeostasis is maintained at the level of absorption (not excretion).

- during Fe deficiency - more iron is absorbed
- during Fe overload - iron absorption decreases
- iron absorption is regulated by the availability of apoferritin in intestinal mucosal cells.
- Fe absorbed in  $Fe^{2+}$  form; then converted to  $Fe^{3+}$   
 $Fe^{3+} + \text{apoferritin} \longrightarrow \text{ferritin}$
- Fe is slowly released from ferritin into plasma as per body requirement.
- As long as iron is in ferritin, Fe absorption is reduced due to reduced availability of apoferritin.
- If transferrin is saturated, any iron accumulation in the form of ferritin in mucosal cells is lost when intestinal cells are desquamated.

## Transport: by transferrin in plasma

- transferrin has 2 binding sites for Fe but can bind only to  $Fe^{3+}$  form  
 $\therefore Fe^{2+} (\text{absorbed}) \xrightarrow{\text{ceruloplasmin}} Fe^{3+}$
- absorbed Fe is mainly delivered to bone marrow for Hb synthesis & further incorporation into developing red cells.

## Iron Utilization: Fe is required by all body tissues

- ↳ required more in bone marrow for Hb synthesis in developing erythrocytes.

## Storage:

**Ferritin:** Fe is stored mainly in liver (96%); spleen, bone marrow, mucosal cells as ferritin

**Hemosiderin:** liver, spleen (mainly in iron excess conditions)

Functions: (Heme = Fe + Protoporphyrin IX)

Heme Proteins:

- for Mb & Mb

- for cytochromes (b, c, c<sub>1</sub>, a<sub>3</sub>, a)
- for cytochrome P450 (in liver; for detoxification)
- Catalases → destroy H<sub>2</sub>O<sub>2</sub>
- Peroxidases in neutrophils help in phagocytosis
- Enzymes → tryptophan pyrrolase, xanthine oxidase.

Non-heme proteins:

- ferritin & hemosiderin → for storage
- transferrin → for transport
- FeS proteins/enzymes → NADH dehydrogenase (Complex I of ETC), succinate dehydrogenase, ribonucleotide reductase
- Aconitase of Krebs Cycle

Iron Excretion: one-way element

- ↳ once it enters the body, it is utilized & reutilized
- very little is excreted (< 1mg/day)
- menstruation is a major cause of loss of iron in premenopausal women.

Iron Deficiency Anaemia:

Causes — decreased iron intake

- lead poisoning (lead decreases Fe absorption)
- lack of absorption (steatorrhea, gastrectomy, achlorhydria)
- increased iron loss (in hook worm infestation, bleeding in urinary/genital tract, peptic ulcer, hemorrhoids, nephrosis)

## Clinical Manifestations -

- Microcytic hypochromic anemia — Low Hb (<12 g/dL) due to decreased Hb synthesis
- Atrophic glossitis
- Oesophageal web
- Dysphagia
- Koilonychia (spoon-shaped nails)
- Pica (craving for starch)
- Apathy (dull & sluggish due to decreased O<sub>2</sub> supply to tissues)
- Fatigue
- Pallor
- Anorexia

Plummer Vinson  
Syndrome

## Iron Toxicity (Hemochromatosis/Hemosiderosis):

1° hemochromatosis — increased iron absorption from intestine

2° hemochromatosis — may be due to increased iron absorption, repeated blood transfusions, thalassemia

Bantu tribes (in Africa) develop hemosiderosis (aka bantu siderosis) due to low phosphate in their staple diet corn [Phosphate favours iron absorption], cooking food & brewing alcohol in iron vessels.

Complications: excessive deposition of hemosiderin in liver, spleen, heart, pancreas (hemosiderosis)

→ deposition of hemosiderin in liver ⇒ liver cirrhosis

→ in pancreas ⇒ death of pancreatic cells ⇒ diabetes

→ bronze colouration of skin

Bronze  
diabetes

# Fluorine (F):

Distribution: teeth & bones as calcium hydroxy fluoro apatite

Sources: drinking water, fish, tea

RDA: Adults  $\Rightarrow$  2-3 mg

$\rightarrow$  safe limit of fluorine in drinking water is 1 ppm

Functions:  $\rightarrow$  Required for proper tooth development  $\Rightarrow$  fluoride gets incorporated into hydroxyapatite to form fluoroapatite

$\Rightarrow$  fluoroapatite (in enamel & dentine) is more resistant to destruction by bacterial acids & plaques, thus preventing dental caries & tooth decay

$\Rightarrow$  fluoride inhibits bacterial enzymes which produce acids that cause dental caries.

$\rightarrow$  Required for proper bone development

$\rightarrow$  fluoride is the inhibitor of enolase  $\therefore$  used during estimation of blood glucose level.

Deficiency: consumption of drinking water with less than 0.5 ppm of fluorine

$\rightarrow$  dental caries

$\rightarrow$  osteoporosis

Excess: fluorosis

Dental fluorosis: intake above 2ppm  $\Rightarrow$  mottling of enamel

$\Rightarrow$  discolouration of teeth (rough, yellow patches)

Skeletal fluorosis: intake above 20ppm  $\Rightarrow$  hypercalcification leading to increased bone density

$\Rightarrow$  ligaments of spine & collagen of bones get calcified

- In advanced stages, ligaments of spine get calcified

$\Downarrow$

stiff spine  $\Rightarrow$  cripples the individual

} genu

} valgum

## Cobalt (Co):

→ Body contains about 1-1 mg cobalt

### Function:

- ↳ cobalt is a constituent of corrin ring system of vit. B12 & their coenzymes
- cobalt activates glycyl-glycine peptidase & ALA synthase.
- stimulates the synthesis of erythropoietin (to promote erythropoiesis).

Deficiency: pernicious anemia (∵ cobalt deficiency causes cobalamin deficiency)

Toxicity: Cause ⇒ overconsumption

- ↳ polycythemia (increased RBCs in blood).

## Iodine (I):

→ body contains  $\Rightarrow$  25 mg ; 20% of this is stored in thyroid glands

Sources: drinking water, sea food, iodized salt

↳ fruits & vegetables grown on seashore

→ regions of high altitudes are deficient in iodine content in food cultivated in this soil.

$\therefore$  people of mountain regions have higher chances of developing iodine deficiency.

$\therefore$  Himalayan region = goiterous belt.

Goitrogens: substances which interfere with iodine utilization

↳ cabbage, cassava, cauliflower.

RDA: Adults : 100-150  $\mu$ g

Pregnancy : 200  $\mu$ g

Absorption: small intestine, skin & lungs

Storage: 80% of body iodine is stored as iodothyroglobulin in thyroid gland.

Functions: Formation of thyroid hormones

↳ synthesis of thyroxine ( $T_4$ ) & triiodothyronine ( $T_3$ ).

→  $T_3$  is functionally more active than  $T_4$ .

Synthesis of thyroid hormones:

• Tyrosine +  $I_2 \xrightarrow{\text{thyroid peroxidase}} \text{MIT (3-moniodotyrosine)}$

• MIT +  $I_2 \xrightarrow{\text{thyroid peroxidase}} \text{DIT (3,5-diiodotyrosine)}$

• DIT + MIT  $\longrightarrow T_3$

• DIT + DIT  $\longrightarrow T_4$

→ synthesis takes place when tyrosine is still a part of thyroglobulin protein.

→  $T_3$  &  $T_4$  are released from thyroglobulin by proteolytic breakdown.

Deficiency:

Simple endemic goiter: deficiency of iodine in food (mainly in geographical areas away from sea-coast)

Iodine deficiency disorders (IDD): hypothyroidism, cretinism, deaf-mutism, goiter, sub-normal intelligence

Excess: Toxic goiter: exophthalmic goiter.

## Selenium (Se):

Sources: sea food, liver, kidney

RDA: 50-100 µg

Functions:

- **Antioxidant property** - Selenium is an integral part of metalloenzyme glutathione peroxidase which destroys  $H_2O_2$  & prevents it from causing oxidative damage of biological membranes

- **Selenium has vit. E sparing action** - vit. E & selenium (both antioxidants) work synergistically to reduce each other's requirement.

- **Constituent of 5'-deiodinase** - for conversion of  $T_4$  to  $T_3$ .

**Selenocysteine** - 21st a.a ; coded by UGA

Keshan disease: endemic cardiomyopathy

↳ Selenium deficiency

→ name derived from ⇒ Keshan province in China, where soil is deficient in Se

Symptoms:

- nausea
- dizziness
- anorexia
- muscle dystrophy
- enlargement of heart ⇒ cardiac failure.

Toxicity (Selenosis): Se is present in metal polishers & anti-rust compounds

∴ people handling these, generally develop selenosis

Symptoms:

- loss of hair
- weight loss
- irritability
- diarrhoea
- garlic odor to breath

## Zinc (Zn):

→ total body Zn = 2mg

→ prostate gland is particularly rich in Zn.

RDA: 10-15 mg

Source: meat, milk, eggs, shellfish

Absorption: mainly absorbed in duodenum

→ metallothionein facilitates Zn absorption

→  $\bar{a}\cdot\bar{a}$  promote Zn absorption ( $\bar{a}\cdot\bar{a}$  = amino acids)

Excretion: excreted through feces

↳ small amounts are excreted in bile

→ not normally excreted through urine

## Function:

Formation of certain metalloenzymes: carboxy peptidase, RNA polymerase, Carbonic Anhydrase, alcohol dehydrogenase, cytosolic superoxide dismutase (SOD), retinal reductase

Participates in visual cycle: as a component of retinal reductase, it helps in regeneration of rhodopsin

Antioxidant role: as a component of SOD

Heme synthesis: as a component of ALA dehydratase

DNA & RNA formation, cell division

Zinc finger motif: → facilitate binding of certain transcription factors with specific regions of DNA

Insulin secretion: Zn is required for storage & secretion of Zn from pancreas

Immunity: Zn maintains integrity of immune system

Taste sensation: Zn containing protein gustin (in saliva) plays an imp. role in taste

Wound healing

Normal reproduction: Zn is required for sexual maturation.

## Deficiency:

Causes: common in children of developing countries due to lack of consumption of non-vegetarian food, high phytate content, diarrhea.

→ Zn deficiency of pregnant women  $\Rightarrow$  spontaneous abortion & congenital malformations.

Symptoms: poor wound healing

- anemia
- loss of taste sensation
- hypogonadism
- diarrhea
- growth failure

Acerodermatitis enteropathica: autosomal recessive condition

↳ defective Zn absorption

→ acerodermatitis is inflammation around mouth, fingers, nose, etc

→ diarrhea

→ ophthalmologic disorders

→ hypogonadism.

Toxicity: seen in welders due to inhalation of Zn-oxide fumes

Symptoms - nausea

- gastric ulcers
- pancreatitis
- pulmonary fibrosis
- vomiting
- anemia
- excessive salivation.

## Molybdenum (Mo):

→ body contains about 9mg

ADA: 45 µg

Source: rich in cereals, legumes, green leafy vegetables, meat

Functions - imp. role in RBC synthesis

- component of many enzymes ⇒ xanthine oxidase, sulfite oxidase, aldehyde oxidase, nitrate reductase

- Mo works with riboflavin to incorporate iron into Hb

Deficiency: Rare

↳ neurological symptoms - mental retardation

→ mouth & oesophageal cancer.

## Manganese (Mn):

→ body manganese = 15 mg

Sources: cereals, nuts, fruits, tea

RDA: 5-6 mg

Functions: as an activator for enzymes — mitochondrial SOD, pyruvate carboxylase, arginase, squalene synthase

- Gluconeogenesis — pyruvate carboxylase
- Fatty acid synthesis — acetyl CoA carboxylase
- Glycoprotein & chondroitin sulphate synthesis — glycosyl transferase
- Cholesterol synthesis — squalene synthase
- Inhibition of lipid peroxidation — SOD
- RNA polymerase activity

Deficiency: rare

↳ reported in case of protein-energy malnutrition (PEM), diabetes → pancreatic insufficiency

→ impaired growth, skeletal deformities

→ defective organic matrix of bone & cartilage

→ raised serum ALP levels

→ abnormal reproductive functions

Toxicity: overconsumption is unlikely

↳ can be caused on prolonged exposure to Mn dust in miners

→ excess Mn accumulates in liver & CNS.

→ severe neuromuscular symptoms (similar to those of Parkinson's disease).